



What can we do to remove PFAS from our drinking water sources?

TEAM 3: PFAS REMOVAL PERFORMANCE TESTING

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Team 3 Investigators



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Research Questions Addressed by Team 3

- What is the best option to remove PFAS from drinking water among commercially available materials, such as activated carbons (AC), ion exchange (IX) resins, and membrane filters?
- What do we do with the waste streams containing PFAS?
- Are there promising novel PFAS removal methods we can develop?
- How successful are the household filters in removing PFAS from tap water?



Preliminary Results



PFAS Removal by High-Pressure Membranes

The quantity and scope of studies evaluating PFAS rejection by high-pressure membranes is limited.

Motivation

Evaluate the impacts of membrane type, PFAS physico-chemical characteristics, and membrane fabrication modifications on PFAS removal by high-pressure membranes

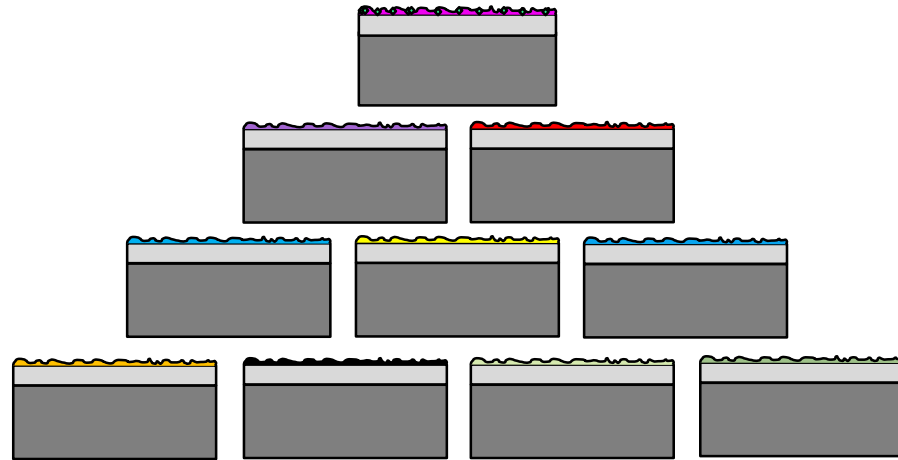
Objective

Experimental Design

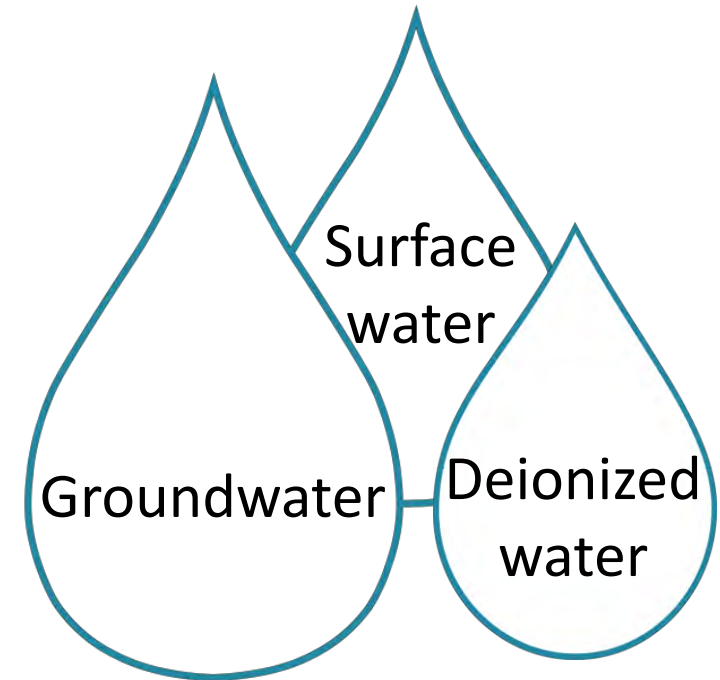
29 types of
PFASs



10 types of
high-pressure
membranes



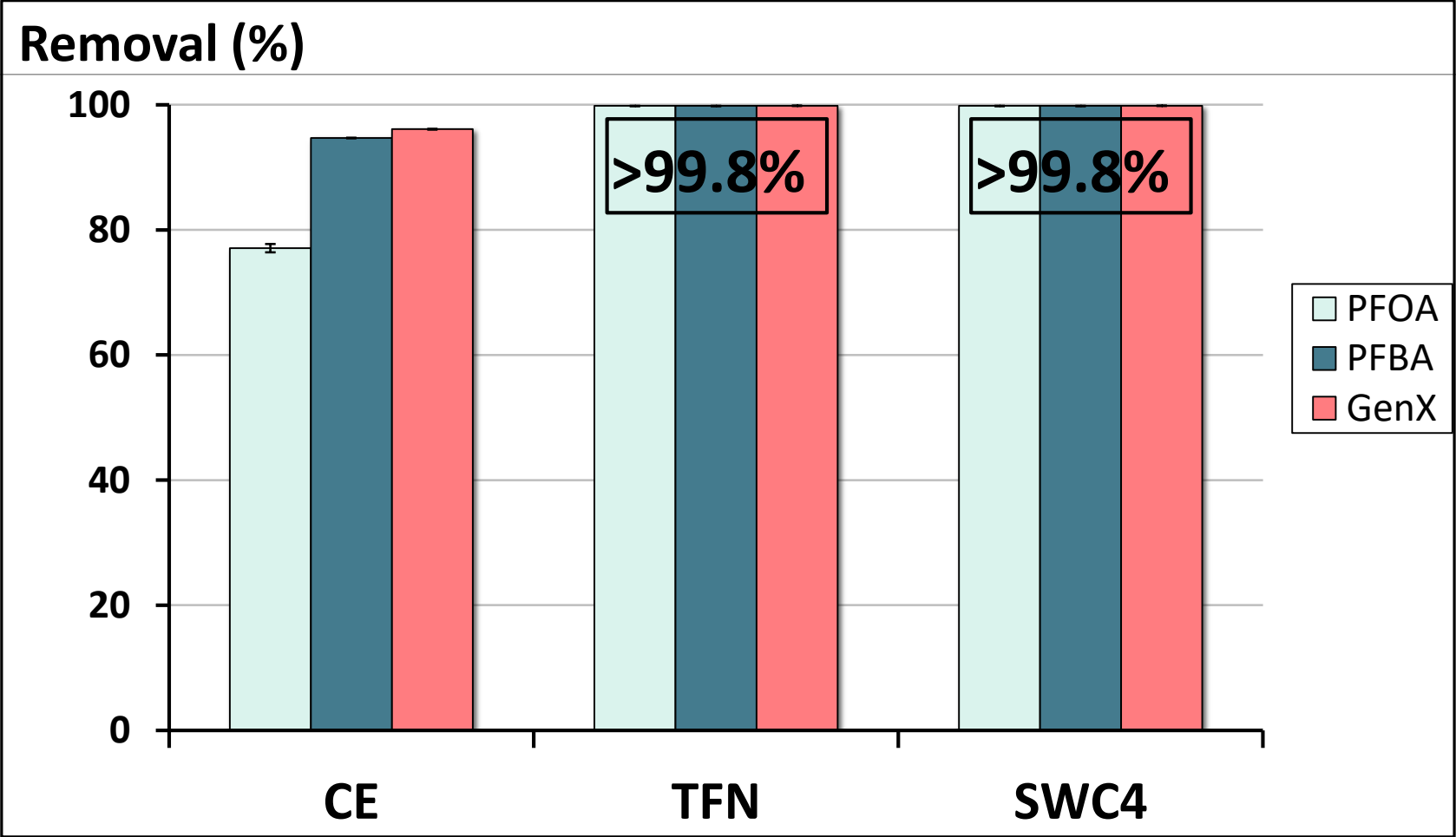
3 types of
water



Preliminary Results



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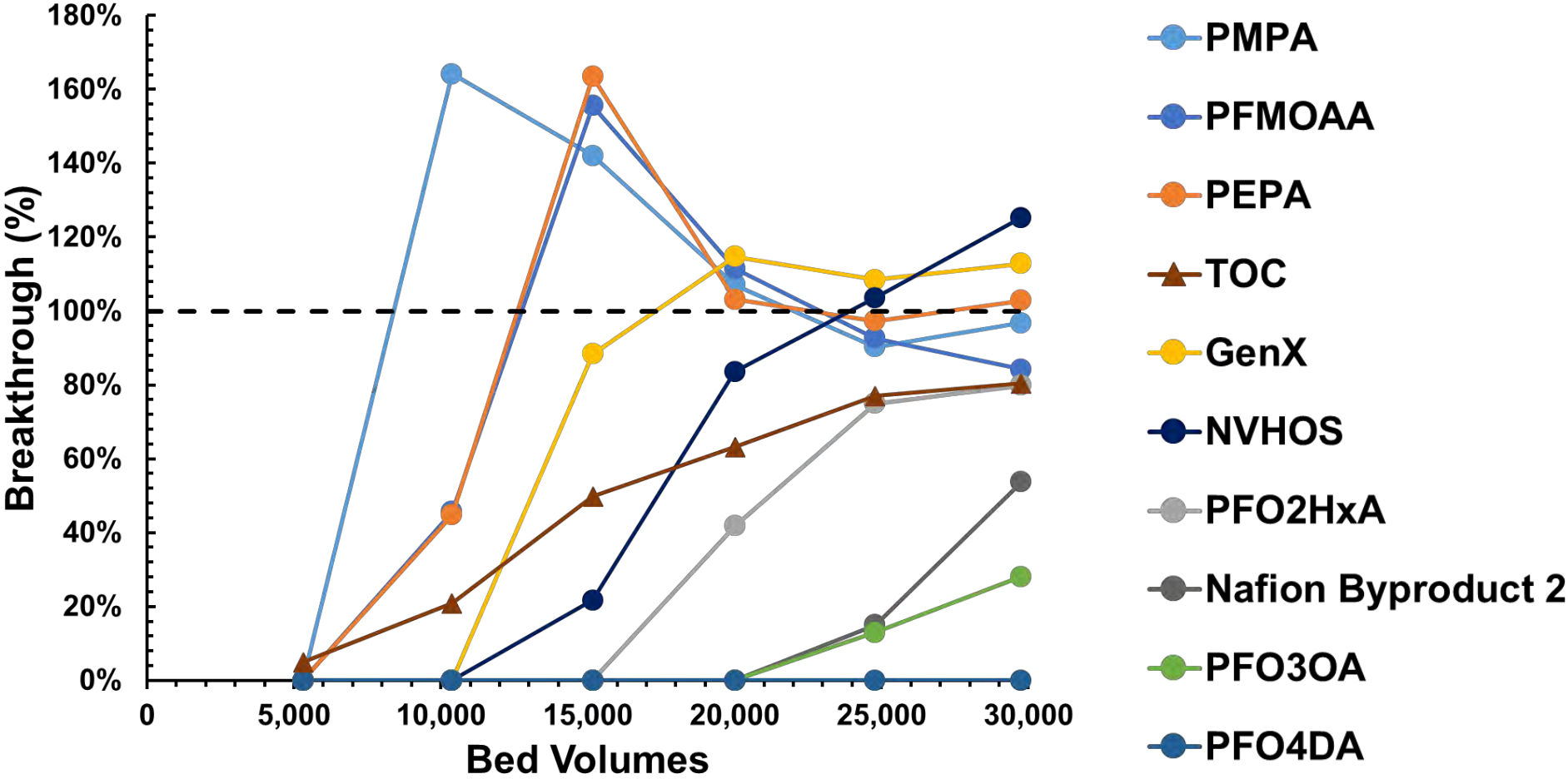
3 types of PFAS

3 types of membranes

3 types of membranes

Effectiveness of granular activated carbon for PFAS removal increases with increasing PFAS chain length

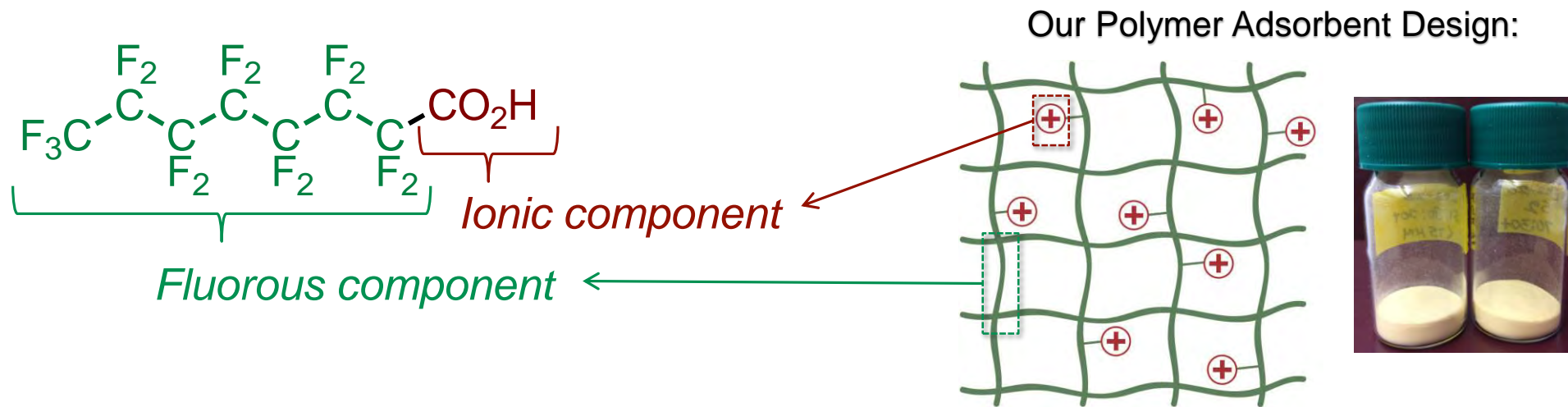
Rapid small-scale column test:



Zack Hopkins, PhD student
CCEE, NCSU

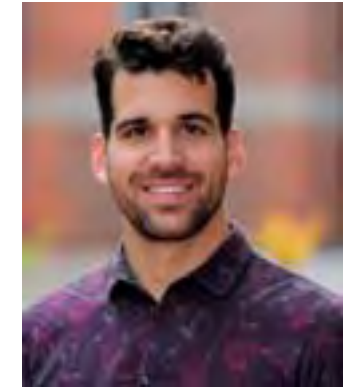
IONIC FLUOROGELS FOR PFAS ADSORPTION

Ionic Fluorogels combine both adsorption ion exchange capabilities to create a sorbent that is selective for PFAS over other organic contaminants.

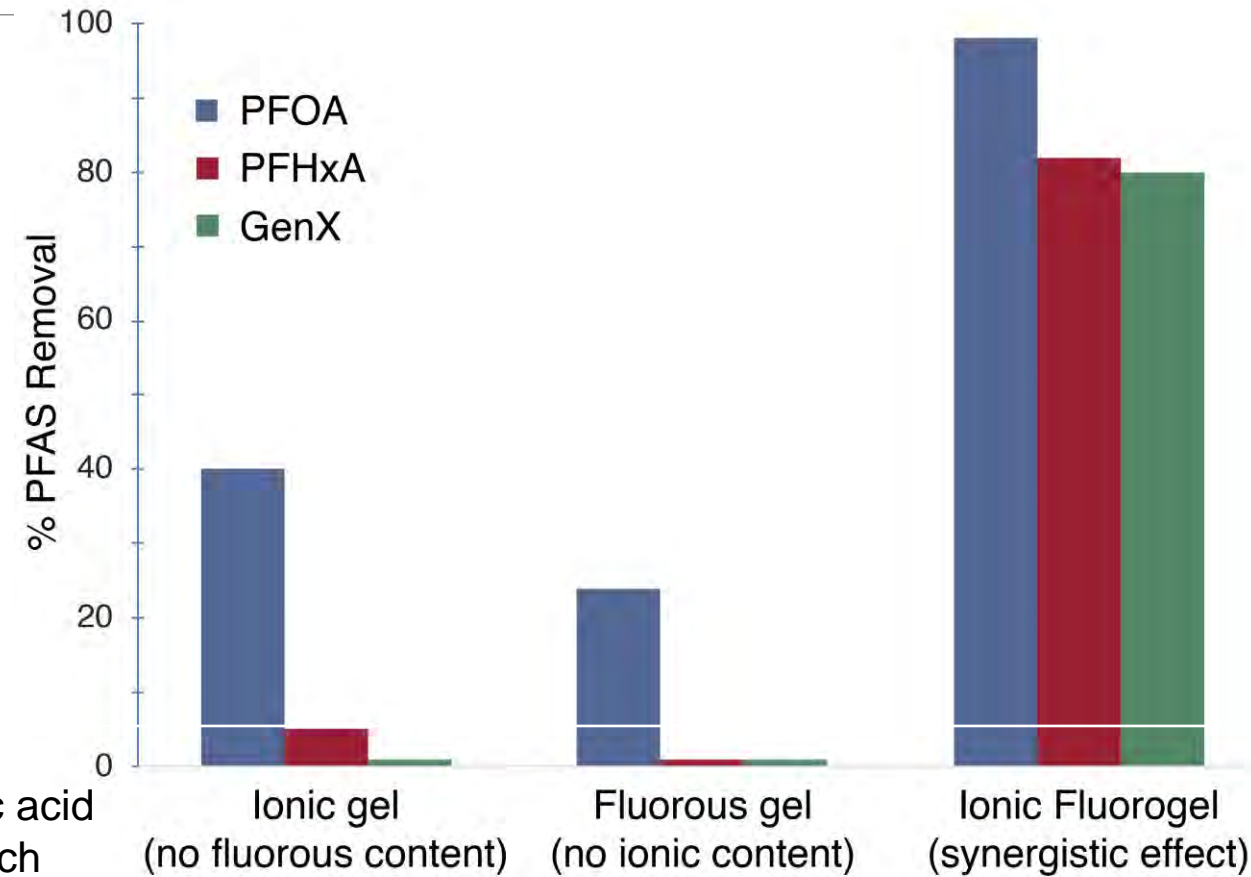


The **Synergistic Combination** of ionic and fluorous components within the granular Ionic Fluorogel resin result in a high capacity and selective PFAS absorbent

GENX ADSORPTION AT ENVIRONMENTALLY RELEVANT CONCENTRATIONS



Frank Leibfarth, PhD
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UNC, Chapel Hill



Conditions:

Challenge: 200 ppm NaCl + 20 mg/L Humic acid

Ionic Fluorogel: 10 mg/L ; PFAS : 1 μ g/L each

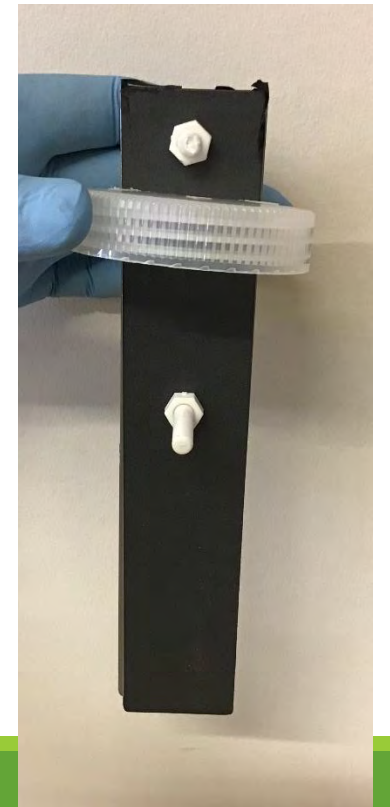
Equilibrium adsorption after 21 h

Average of 3 trials

Electrochemical Mineralization of PFAS

Electrode materials tested

- Ruthenium oxide coated Titanium (Ti/RuO₂)
- Ebonex Plus (a commercial monolithic Ti₄O₇ ceramic material)
- Graphene membrane



Electrochemical Mineralization of PFAS



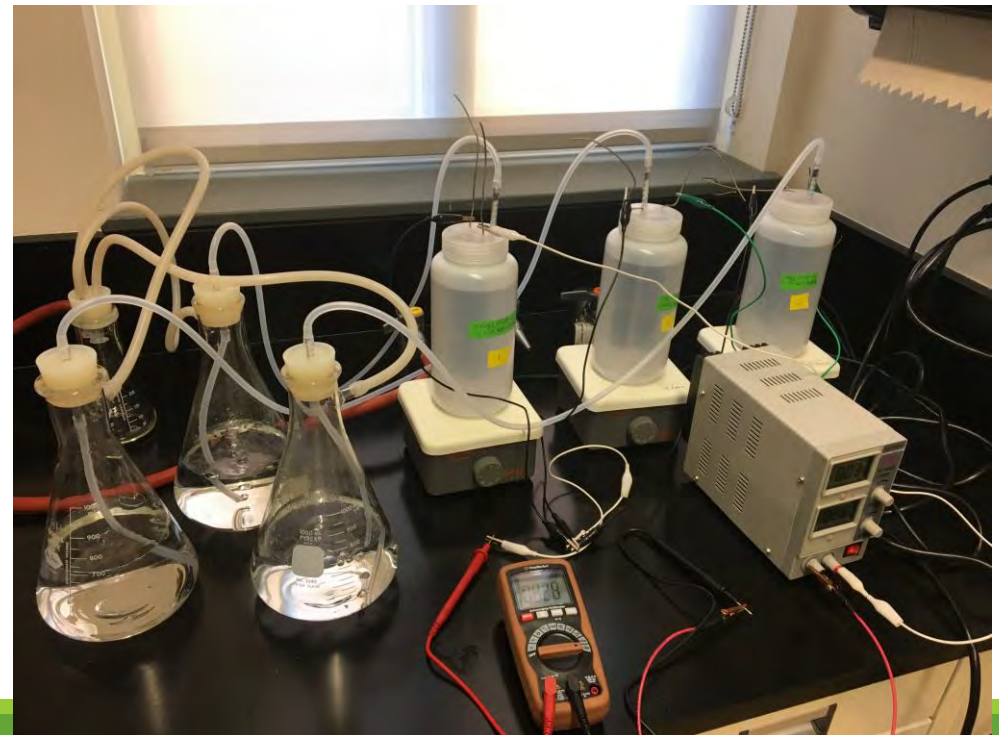
Mei Sun, PhD
Assistant Professor
UNC, Charlotte

Preliminary results:

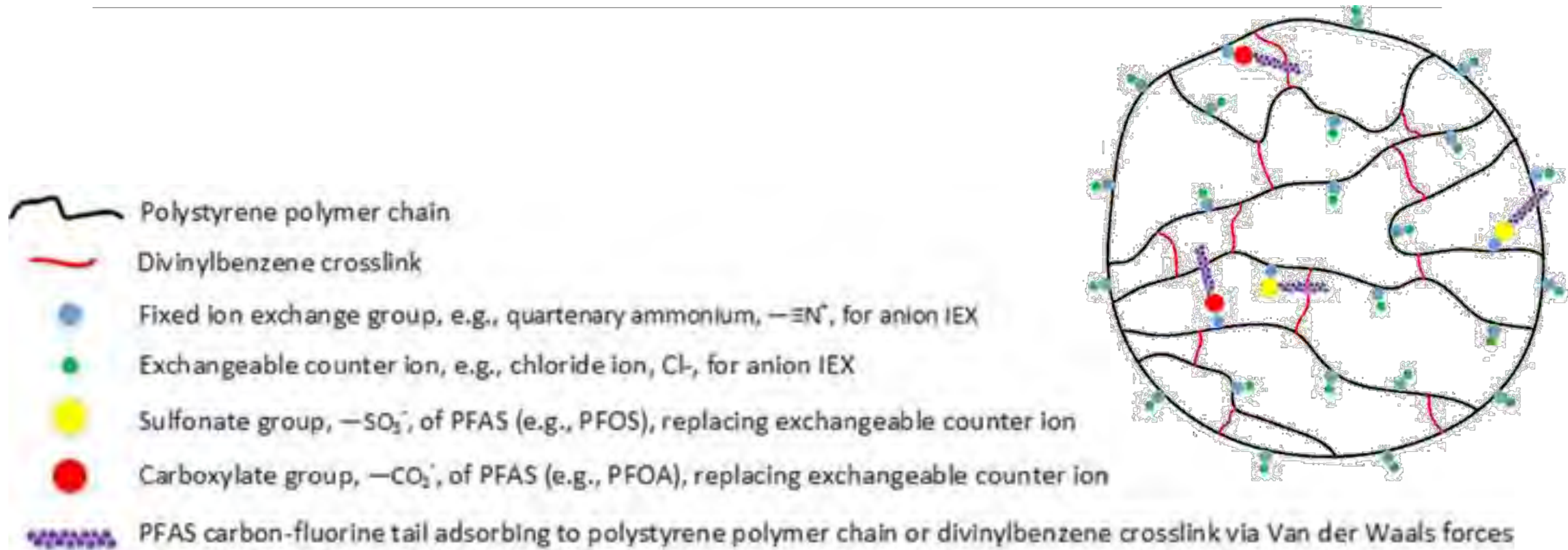
- 93% PFOA removal achieved using the Ti/RuO₂ electrode at 30 mA/cm²

Ongoing work

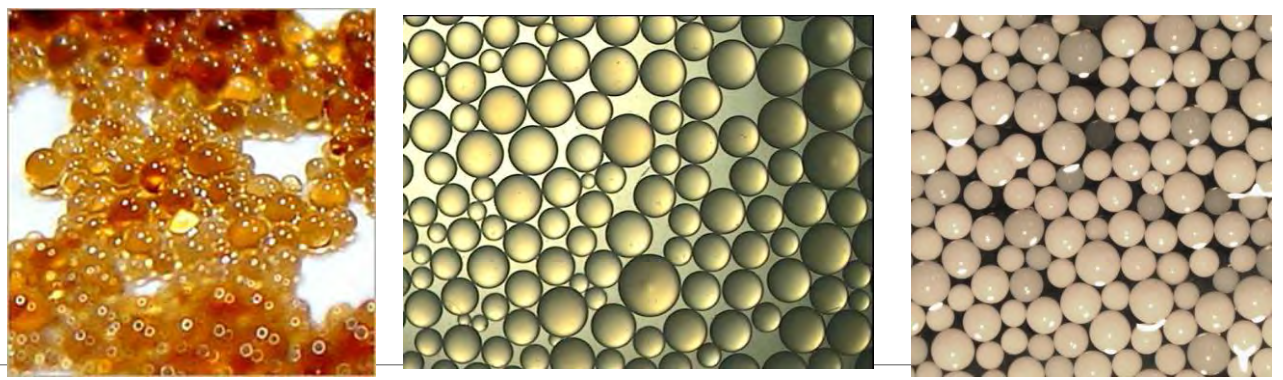
- Identify degradation products
- Test degradation at lower current densities
- Test degradation of other PFAS



PFAS Removal by Ion Exchange (IX)



Research Plan



Screening tests for the most effective resins

- 5 DOW resins, 3 Purolite resins and 2 IXOM resins based on literature review

Water matrices

- groundwater from Fayetteville
- Surface water from Wilmington

Ongoing and future studies

- PFAS removal efficiency
- PFAS removal kinetics
- PFAS sorption isotherms
- Effects of other constituents in water on PFAS removal
- Breakthrough curves
- Resin regeneration

Preliminary Results: PFAS removal by In-Home Water Filters



Nicholas Herkert, PhD
Post-doc; Duke University



Refrigerator Filter



Pitcher Filter



Reverse osmosis

Analyte	Refrigerator Filter	Pitcher Filter	Reverse Osmosis Filter
GenX	56%	46%	100%
PFBA (4 carbon)	47%	36%	100%
PFHxA (6 carbon)	60%	43%	100%
PFOA (8 carbon)	73%	69%	100%

Expected Timeline and Products



- Targeted date to complete research: January 2020
- Expect to produce several research reports with information on:
 - Recommendations on the types of materials to use in large scale water treatment plants to optimize the removal of PFASs in NC waters
 - Information on approaches for degrading PFASs from waste streams
 - Recommendations for home water filtration systems to remove legacy and emerging PFASs in NC drinking water
- Communicate and discuss findings with relevant stakeholders in 2020
 - NC Legislature
 - NC DEQ
 - Water utilities

Thank you for your attention!

Questions?

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